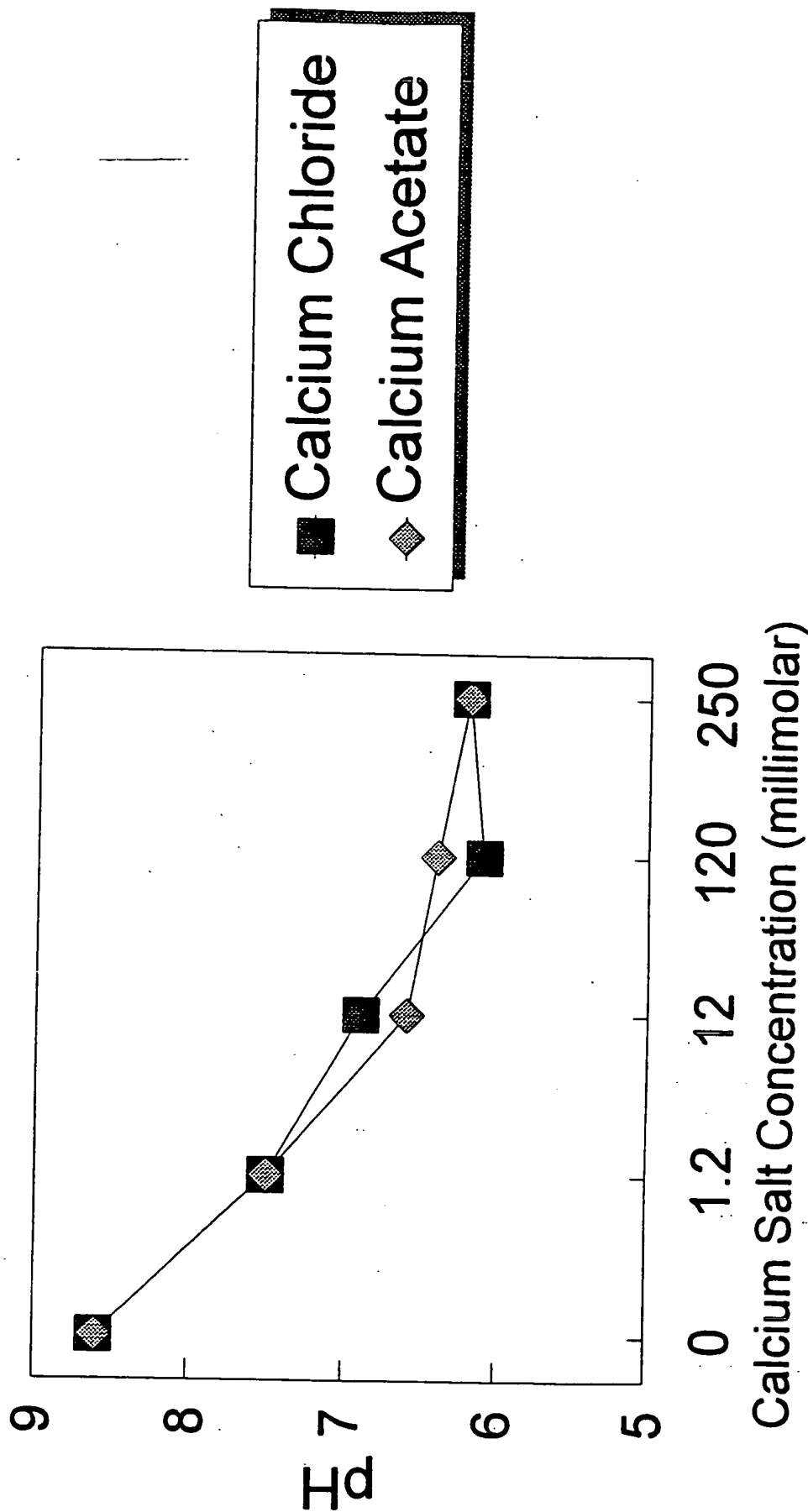


**Figure 1. The Effect of Calcium Chloride and Calcium Acetate
on the pH of a Calcium Carbonate Slurry**

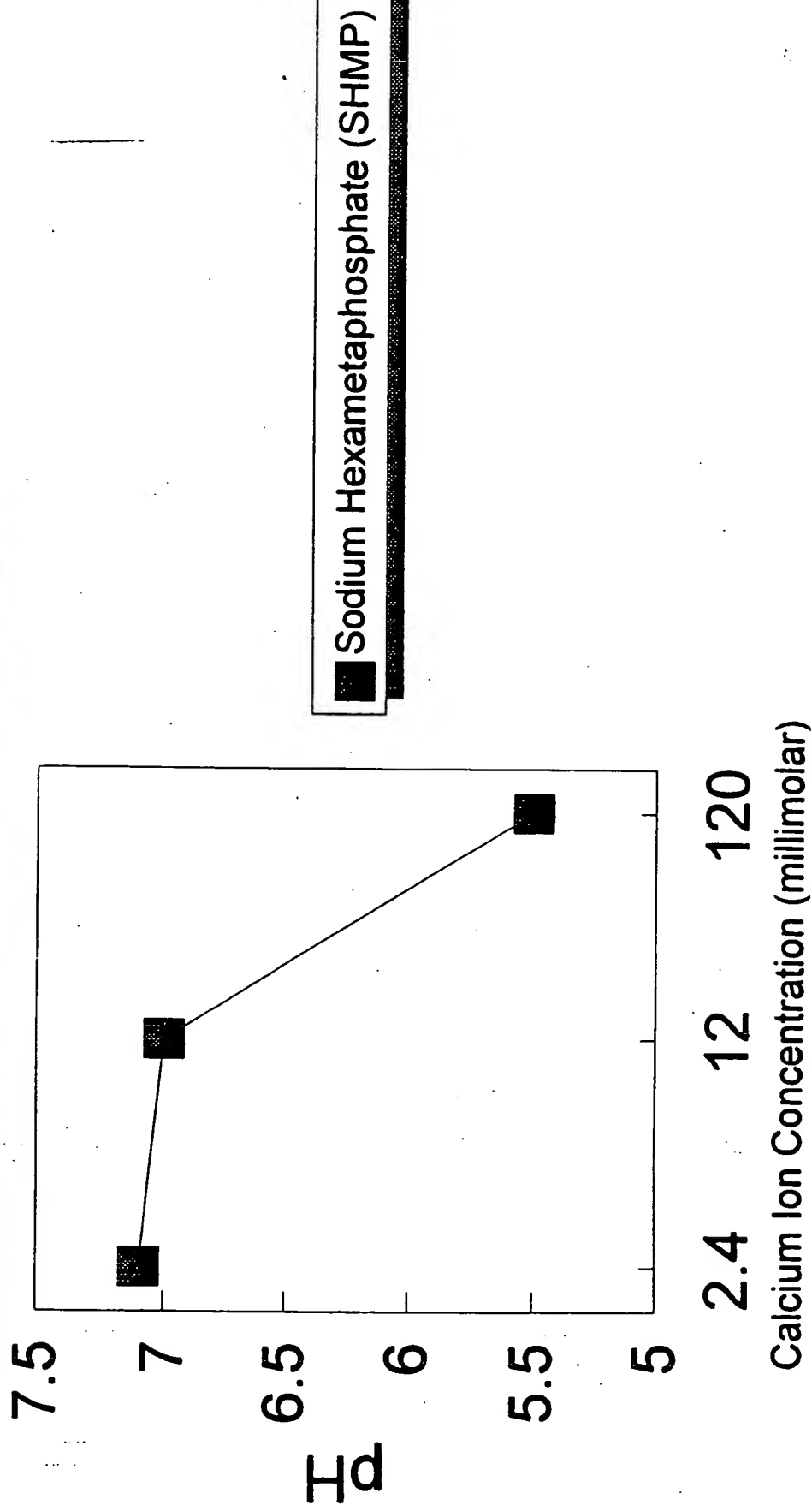


**The slurry contains 5% calcium carbonate and
the pH was measured after 4 days.**

Time (days)	HD 10106 ($W_{H\alpha}$ in Å)	HD 10107 ($W_{H\alpha}$ in Å)	HD 10108 ($W_{H\alpha}$ in Å)
0	~6.00	~5.80	~5.40
1	~6.05	~5.78	~5.42
2	~6.08	~5.75	~5.45
4	~6.10	~5.72	~5.48
10	~6.12	~5.70	~5.50

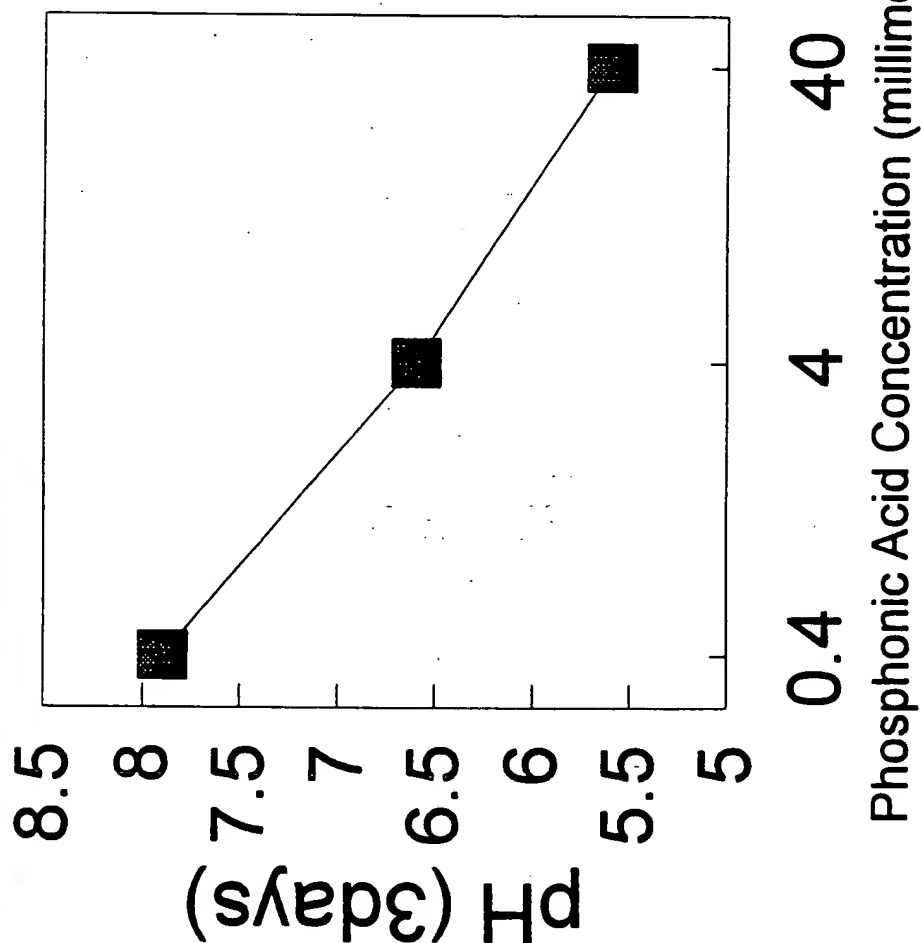
The slurry contains 5% precipitated calcium carbonate under one atmosphere of carbon dioxide.

**Figure 3. The Effect of Calcium Ion Plus Chelate
on the pH of a Calcium Carbonate Slurry**



The slurry contained 5% calcium carbonate and 0.7 millimolar SHMP and the pH was measured after 3 days.

Figure 4. The Effect of Phosphonic Acids on the pH of a Calcium Carbonate Slurry



■ Phosphonic Acid

The slurry contained 5% calcium carbonate and the acid was Nitrilotri(methylene)triphosphonic Acid